

# Coupling Pb and Zn bioaccessibility with sequential and HNO<sub>3</sub> extraction in soil from the industrial area of Volos, Greece

KELEPERTZIS E.<sup>1</sup>, BOTSOU F.<sup>1</sup>, SUNGUR A.<sup>2</sup>, SIGALA E.<sup>1</sup>, DAFERERA O.<sup>1</sup>, KYPRITIDOU Z.<sup>1</sup>,  
CHRASNÝ V.<sup>3</sup>, ARGYRAKI A.<sup>1</sup>, KOMÁREK M.<sup>3</sup>, SKORDAS K.<sup>4</sup>

<sup>1</sup> National and Kapodistrian University of Athens, Panepistimiopolis, Zografou 15784, Athens, Greece

<sup>2</sup> Çanakkale Onsekiz Mart University, Çanakkale 17020, Turkey

<sup>3</sup> Czech University of Life Sciences, Prague-Suchbát 165 00, Czech Republic

<sup>4</sup> University of Thessaly, Volos, Greece

\*corresponding author: Efstratios Kelepertzis

e-mail: kelepert@geol.uoa.gr

**Abstract** The urban environment is a complicated system where various anthropogenic sources contribute to the accumulation of metals in soil, leading to potential negative environmental effects. Within this context, we performed single and sequential extractions in contaminated soil from the industrial area of Volos. The objective was to understand how Pb and Zn oral bioaccessibility was related to different pools in soil based on sequential and 0.43 M HNO<sub>3</sub> extractions. Pb was principally found in the reducible fraction (F<sub>2</sub>; 15-76%), followed by the residual (F<sub>4</sub>; 9-54%) and the oxidizable (F<sub>3</sub>; 6-54%), whereas the acid soluble fraction was of minor importance (F<sub>1</sub>; 2-15%). Zn fractionation was dominated by the residual fraction (20-73%), followed by the oxidizable (10-49%) and the reducible (5-38%). Significant correlations were found between Pb and Zn bioaccessibility and the sum of fractions F<sub>1</sub>, F<sub>2</sub> and F<sub>3</sub> (0.50 for Pb and 0.86 for Zn,  $p < 0.01$ ), indicating that the applied bioaccessibility solution preferentially targeted the mobile fraction of Pb and Zn in soil. HNO<sub>3</sub>-extractable Pb and Zn were highly correlated to bioaccessible Pb and Zn (0.80 and 0.93 respectively,  $p < 0.01$ ), suggesting that the HNO<sub>3</sub> can determine the oral bioaccessibility of Pb and Zn in urban soil influenced by industrial activities.

**Keywords:** soil contamination, toxic elements, geochemical fractionation, bioavailability

## 1. Introduction

Soil contamination in urban and industrial areas is a significant environmental problem worldwide (Wong et al., 2006). In such environments, potentially toxic elements (PTEs) like Pb and Zn, have been used as tracers of anthropogenic contamination. PTEs enrichment in industrial and urban soil has been well-documented, originating from industrial activities, coal combustion and traffic emissions (Luo et al., 2012; Wei and Yang, 2010).

Risk assessment based on total PTEs content may overestimate the actual risk for the environment. These elements in soil are bound to different geochemical

fractions, such as the acid soluble and exchangeable, reducible (Fe and Mn oxides), oxidizable (organically bound) and residual (bound to primary minerals), with different mobility and solubility characteristics (Rao et al., 2008). For human health risk assessment, the oral bioaccessibility of an element is defined as the fraction of the element that is soluble in the gastrointestinal environment, representing the maximum amount of contaminant that is available for absorption. In addition, the 0.43 M HNO<sub>3</sub> extraction has been used to extract the reactive fraction of PTEs, which are sorbed onto amorphous metal oxides (Römken et al., 2009). Recently, it has been suggested that this specific single extraction may be used as an alternative for PTEs bioaccessibility (Rodrigues et al., 2018). Nonetheless, limited studies have focused on the relationship between PTEs in bioaccessible and reactive pools, and PTEs in different geochemical fractions.

Volos is a medium-sized city, with approximately 150,000 inhabitants. Heavy industrial activities include the operation of one cement production plant about 3 km east of the city center, and a major steel making plant, located some 20 km west of the city. In our previous studies, the soil contamination by PTEs has been addressed (Kelepertzis et al., 2020; Kelepertzis et al., 2021). However, the geochemical partitioning of PTEs has not been studied yet, neither has the evaluation of their reactive fraction. We focus on this study on Pb and Zn because these metals can be considered as indicative of local industrial activities, mostly related to the operation of the steel plant.

The objectives of this study were to (a) determine Pb and Zn fractions contributing to bioaccessible Pb and Zn in soil from the industrial area of Volos, and (b) compare Pb and Zn bioaccessibility to their reactive pools with the aim to evaluate if the 0.43 M HNO<sub>3</sub> extraction is an alternative for Pb and Zn oral bioaccessibility.

## 2. Materials and methods

### 2.1. Sampling and samples pre-treatment

A total number of 29 soils (0-10 cm depth) were selected from the sample data set of an earlier survey in the wider area of Volos (Kelepertzis et al., 2020). The criteria for sample selection were the total content of Pb and Zn as determined by a strong acid dissolution. These samples were located around the steel mills ( $n=7$ ), the cement plant ( $n=10$ ), as well as within the city core ( $n=12$ ) (Fig. 1). The land use of the wider area and the rocks outcropping are shown in Fig. 1. Laboratory sample preparation included sieving through 2 mm and subsequently 100  $\mu\text{m}$  nylon sieves to focus on geochemically reactive and bioaccessible particles. Details on sample collection and preparation and analytical methods for the determination of total concentrations of Pb and Zn are described in our previous publication (Kelepertzis et al., 2020).

### 2.2. Laboratory methods for the evaluation of Pb and Zn bioaccessibility, reactivity and geochemical fractionation

The oral bioaccessibility of Pb and Zn was determined by using a 0.4 M glycine extraction solution (adjusted to pH 1.5 with concentrated HCl), simulating the low pH conditions of the human gastric fluids (USEPA, 2013: Method 1340). Details for the analytical procedure of bioaccessibility assessment following the Simplified Bioaccessibility Extraction Test (SBET) can be found in the study of Kelepertzis et al. (2021).

The reactive forms of Pb and Zn were obtained by mixing 1 g of sample with 40 ml of a 0.43 M  $\text{HNO}_3$  solution and shaking for 2 h at room temperature (Rodrigues et al., 2018). The extracts were separated from the solid residue by centrifugation at 3500 rpm for 10 min, and filtration through a 0.45  $\mu\text{m}$  filter.

The soil samples were also subjected to the three-step sequential extraction procedure proposed by the Standards, Measurements and Testing Programme (formerly BCR) of the European Commission (Rauret et al., 1999) (known as modified BCR sequential extraction and named hereafter as BCR). The procedure includes three successive extractions with 0.11 M acetic acid, 0.5 M hydroxylammonium chloride in 0.05 M nitric acid (reducing agent) and 1 M ammonium acetate at pH 2 after digestion with 8.8 M hydrogen peroxide (oxidizing agent). An additional step was included to dissolve elements from the residue remaining after the three extraction steps. The BCR procedure is able to separate elements into geochemical fractions of acid soluble and exchangeable (F1: ion-exchange and carbonate bound), reducible (F2: Fe-Mn bound), oxidizable (F3: organically bound) and residual (F4: primary silicate bound). Details on the BCR sequential extraction method can be found elsewhere (e.g. Rao et al., 2008).

Concentrations of Pb and Zn from the 0.43 M  $\text{HNO}_3$  and the sequential extraction procedures were determined by flame atomic spectrometer. Certified reference materials, procedural blanks and analytical duplicates were used for quality control purposes.

### 2.3. Statistical analysis

All variables were screened for normality of their distribution by the Shapiro-wilk test. Because of violations of normality, the Spearman rho coefficient was used to explore relationships between bioaccessible, reactive and geochemical fractions of Pb and Zn. Plotting of geochemical data was performed with OriginPro 2016 (OriginLab Corp.)

## 3. Results and Discussion

The percentage bioaccessible and reactive fraction was calculated as follows:

$$\% \text{ bioaccessibility (or reactivity)} = \frac{C_{\text{bio/react}}}{C_{\text{tot}}} \times 100$$

where  $C_{\text{bio}}$  or  $C_{\text{react}}$  is the bioaccessible or reactive concentration of Pb and Zn (in mg/kg), respectively, and  $C_{\text{total}}$  is the total content of Pb and Zn (in mg/kg), determined by a strong acid ( $\text{HNO}_3$ - $\text{HClO}_4$ -HF) dissolution (Kelepertzis et al., 2020).

The median bioaccessibility was 36% for Pb (range from 14% to 69%) and 37% for Zn (range from 11% to 65%) (Fig. 2a). Considering the reactivity, median Pb and Zn  $\text{HNO}_3$ -extracted ratios were 58% (range from 28% to 80%) and 37% (range from 15% to 76%), respectively (Fig. 2b). Such Zn bioaccessible and reactive ratios are comparable to those reported for the highly urbanized city of Athens (Kelepertzis and Argyraki, 2015). Nevertheless, Pb reactive and bioaccessible data for Athens soil are higher than those measured in the soil from the industrial area of Volos. Since the pH conditions during both the extraction tests are similar, the observed differences in the metal bioaccessibility and reactivity ratios should be due to the variety of geochemical forms defining the speciation of Pb and Zn in soil.

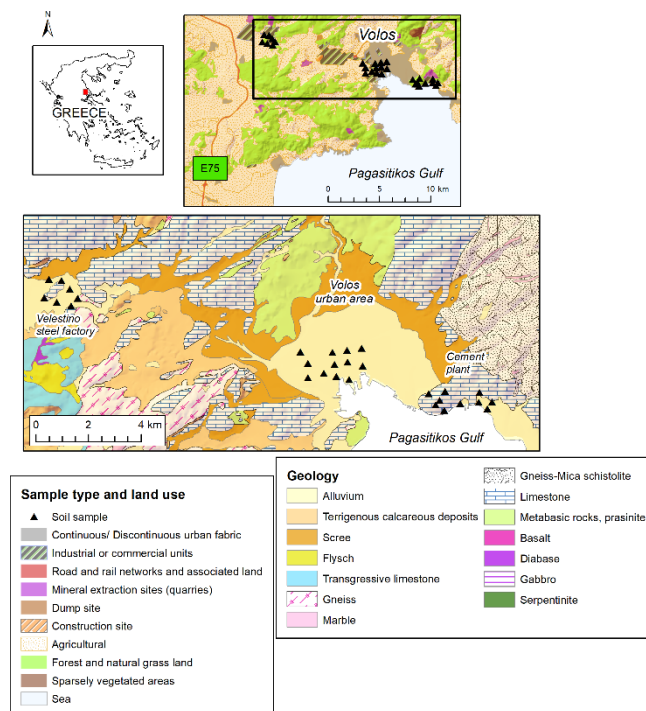
To understand how different geochemical phases contribute to Pb and Zn bioaccessibility, the soil samples were separated into 4 fractions via the sequential extraction procedure (Fig. 2c). The most important fraction of Pb based on medians was the reducible (51%), followed by the residual and the oxidizable (21% and 20%, respectively), whereas the acid soluble fraction comprised only 8% of total Pb content. In the case of Zn, the order of its fractionation was: residual (40%) > oxidizable (24%) > reducible (23%) > acid soluble (8%). The sum of fractions F1, F2 and F3 is considered to represent the potential mobile fraction of elements under natural conditions. The mobile fractions of Pb and Zn accounted for 79% (range 46 to 91%) and 60% (range 27-80%), respectively.

To explore how each Pb and Zn fraction contributed to the bioaccessible Pb and Zn, we tested the correlations between Pb and Zn in the different fractions, as well as in the mobile one, with bioaccessible Pb and Zn. For both elements, statistically significant Spearman rho correlations were found between the bioaccessible and the sum of the first three fractions of the BCR (0.50 for Pb and 0.86 for Zn,  $p < 0.01$ ) (Fig. 3), indicating that bioaccessibility (%) is governed by the mobile pools of Pb and Zn present in the soil, *i.e.* mostly bound to Fe and Mn oxides and organic matter. Interestingly, no correlations were found between the bioaccessibility of Pb and the each

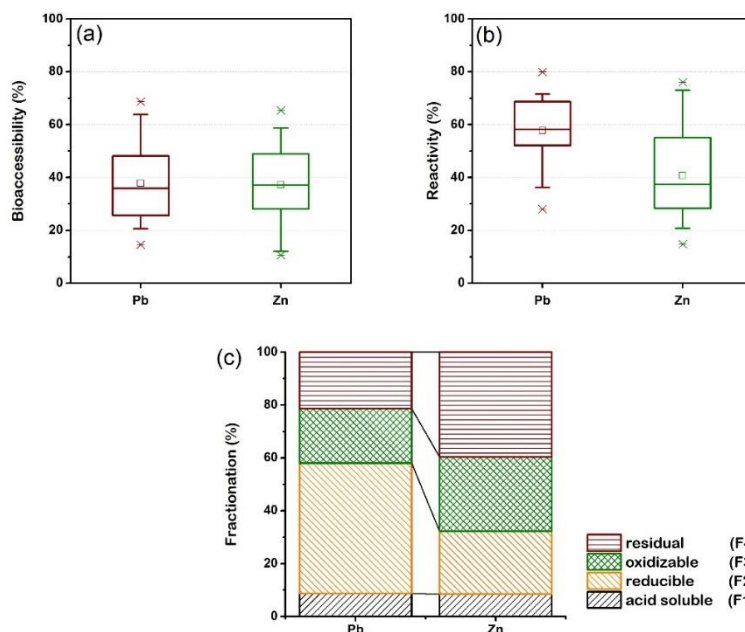
of the F1, F2 and F3 fractions. Regarding Zn, correlations between each Zn fraction and respective bioaccessibility exist, showing that the applied bioaccessibility solution dissolves Zn bound to the first three phases of the BCR.

Moreover, HNO<sub>3</sub>-extractable Pb and Zn were correlated with Pb and Zn bioaccessibility (Fig. 3), with correlation coefficients being 0.80 for Pb and 0.93 for Zn (p<0.05). This was consistent with results published by Kelepertzis et al. (2015) and Rodrigues et al. (2013) who demonstrated

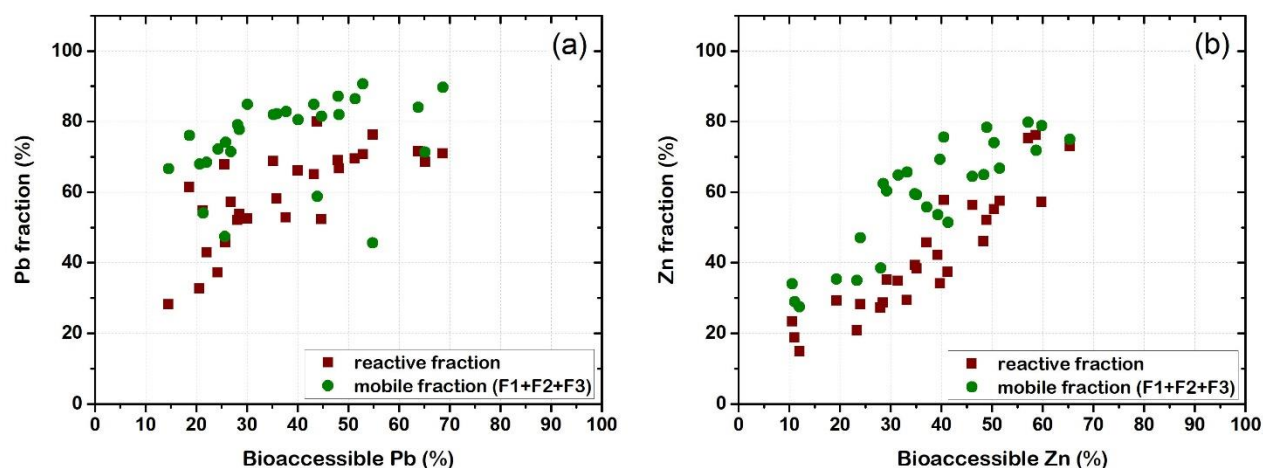
strong correlations between the two methods for a considerable number of soils from Athens and Porto, respectively. One reasonable explanation for the close relationship is the similarity in the extraction pH that is close to 1 in both solutions (HNO<sub>3</sub>: pH ~ 1, SBET: pH = 1.5) (Li et al., 2015). These results demonstrate that a single extraction with dilute HNO<sub>3</sub> can be used to determine the bioaccessibility of Pb and Zn in urban soil influenced by industrial emissions.



**Figure 1.** Sampling locations plotted on land use and geological map (Katsikatsos et al., 1978) of the study area.



**Figure 2.** Geochemical results of Pb and Zn based on single and sequential extractions for the soil samples from the wider area of Volos: a) bioaccessibility (%), b) reactivity (%), c) geochemical fractionation (%)



**Figure 3.** Relationships between bioaccessible Pb and Zn and their reactive fraction, as well as the sum of the first three fractions (F1, F2, F3) from the sequential extraction procedure

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